

Grand Challenge

Grand Challenges in CO₂ Capture and Conversion

Maedeh Kafi, Hamidreza Sanaeepur*, Abtin Ebadi Amooghin*

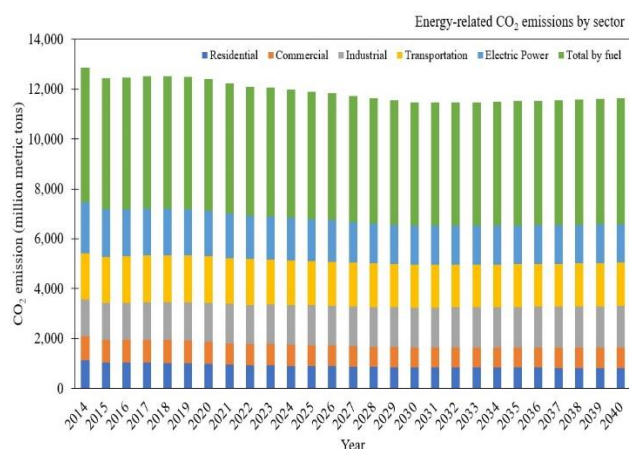
Department of Chemical Engineering, Faculty of Engineering, Arak University, Arak 38156-8-8349, Iran

ABSTRACT

CO₂ and its emission control is one of the main challenges in climate change mitigation. There are various methods for CO₂ capture, including physical and chemical technologies such as chemical looping, post-combustion, pre-combustion, reduction and bio-technologies. Besides these methods, there are methods to convert CO₂ into value-added products. However, both approaches face challenges that limit their commercialization. In this paper, the challenges of CO₂ capture and conversion are examined and pros and cons of the methods to remove these obstacles are studied. Here, as a result, four main challenges in CO₂ capture and conversion were presented: (1) energy consumption of existing technologies and some alternatives, (2) fixed and operational costs, (3) insufficient activity, sustainability and economics of existing catalysts or microorganisms for CO₂ utilization and conversion, and (4) carbon footprint in existing technologies. Also, it was concluded that the need for more reliable life cycle assessment data for zero carbon footprint in existing and future CO₂ capture and conversion technologies is one of the most important concerns that should be addressed in future studies to explore creative solutions for this issue.

KEYWORDS: CO₂ capture and conversion; Climate change; Energy; Costs.

GRAPHICAL ABSTRACT



HIGHLIGHTS

- CO₂ as a major challenge in climate change.
- CO₂ capture and conversion into value-added products.
- Energy, costs, effective catalysts, and carbon footprint as four main challenging issues.

1. Introduction

Since the era of industrialization, the amount of carbon dioxide (CO₂) emissions has increased and has gradually aggravated climate change (Burkart et al., 2019; Gao et al., 2020). Most CO₂ emissions come from the combustion of fossil fuels and can therefore be reduced through the gradual elimination of fossil fuels. However, although fuel eliminations might be possible for the power industry and transportation sector (through substitution with renewable

energies), many sectors (e.g., steel/cement production, intercontinental air transport or non-electrical trains) do not have a suitable alternative for carbon-based fuels (Figure 1). Therefore, to achieve net zero greenhouse gas (GHG) emissions, in addition to emission reduction, CO₂ capture, and utilization are required in the long term (Castel et al., 2021; Sharifian et al., 2021).

Correspondence to: H. Sanaeepur and A. Ebadi Amooghin.
E-mail address: h-sanaeepur@araku.ac.ir and a-ebadi@araku.ac.ir.

<http://doi.org/10.52547/jrr.2302-1007>

Received: 2023-02-27; Revised: 2023-03-28; Accepted: 2023-03-29; Published: xxx
© 2023 Membrane Industry Development Institute. All rights reserved.

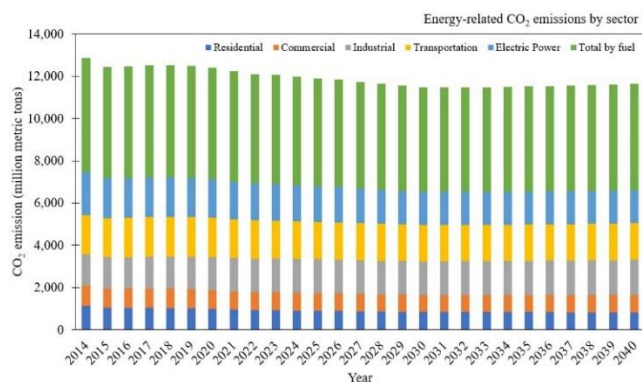


Fig. 1. Energy-related CO₂ emissions by sector (U.S. Energy Information Administration, 2021).

To combat climate change, an important strategy is to close the anthropogenic carbon cycle through carbon capture and utilization (CCU). Therefore, there is a fundamental need to focus on the effective combination of CO₂ capture with further conversion through mineral carbonation, chemical and biological conversion to other chemicals or fuels (Burkart et al., 2019; Gao et al., 2020; Castel et al., 2021). Centralized capture technologies are often grouped into main categories of: (1) chemical looping (combustion), (2) pre-combustion, and (3) post combustion. In addition, CO₂ capture technologies, using algae, biochar, charcoal, and nature-based solutions including tree planting (afforestation and reforestation), where CO₂ is essentially removed from the atmosphere and is converted into biomass through photosynthesis, has recently been investigated (Sharifian et al., 2021; Galán-Martín et al., 2022).

In general, there are four main challenges in CO₂ capture and conversion, including:

1. The main challenge currently limiting CO₂ capture and conversion methods is the high energy consumption. Processes such as adsorption and reduction require high energy for adsorbent regeneration and CO₂ activation, respectively (Wei et al., 2022).

Table 1

CO₂ products and the relating capture methods.

Technology			Production and utilization
CO ₂ conversion			Mineral carbonation (cement and concrete) (Burkart et al., 2019)
Industrial synthesis			Urea and Salicylic acid (Burkart et al., 2019)
Catalytic			C ₁ Products: Carbon monoxide, Methane, Formaldehyde, ... (Burkart et al., 2019)
			Ethylene and Ethanol (Burkart et al., 2019)
			Cyclic carbonates, Polymers (polycarbonates, ...) (Burkart et al., 2019)
			Hydrocarbons (De Ras et al., 2019)
			Liquid fuels (Burkart et al., 2019)
			Higher alcohols (Li et al., 2018)
			Carboxylic acids (acrylic acid, methacrylic acid, ...) (Burkart et al., 2019)
Electrochemically			Oxalic acid (Burkart et al., 2019)
Biological	Photosynthetic	Green algae	Fuels (biodiesel, renewable diesel/gasoline), Protein, Pigment (Burkart et al., 2019)
		Cyanobacteria	Fuels (ethanol, iso-butanol, n-butanol), Fatty acids, Ethylene, ... (Burkart et al., 2019)
	Non-photosynthetic	Chemo-lithotrophic organisms	Acetogens (Burkart et al., 2019)
		Bioelectrochemical organisms	Succinate, Acetate, Alcohols, Electricity, ... (Burkart et al., 2019)
	Biorefinery (indirect)		Biofuels, Bio-materials, Electricity, Biochar, Green hydrogen, ... (Galán-Martín et al., 2022)
Direct utilization of CO ₂			Refrigerant, Fire extinguishers, Carbonated beverages, ... (Burkart et al., 2019)
			An excellent working fluid and solvent for many applications such as: power generation, dyeing agent, and food-safe chemical extractions (Valluri et al., 2022)

2. The second challenge is the cost of direct air capture (DAC) through traditional technologies (ranging from \$100 to \$1000 per tonne of captured CO₂) (Sharifian et al., 2021).

3. The development of highly active, stable, and economical catalysts or microorganisms for CO₂ utilization is crucial, although difficult and challenging, for practical systems that reduce CO₂ emissions (Burkart et al., 2019).

4. CCU's carbon footprint is also an issue (Castel et al., 2021). Carbone neutralizing scenarios are needed in this case. First, advanced separation and sequestration technologies in the short-term development can be developed to mitigate the climate change impact of CCU technologies. Second, substitution of renewable energies or development of fuels or chemicals with equivalent energies and zero emissions instead of the present fossil fuel-based ones is a better scenario that can be considered in the sustainable perspective (Centi and Parathoner, 2023; Chung et al., 2022).

After the CO₂ is captured, it can be stored or utilized. Table 1 shows the CO₂ products and the relating capture methods. CO₂ is an inexpensive, non-toxic, renewable commodity. The CO₂ utilization market is projected to grow from 0.23 gigatonnes per year (Gt/year) today to 7 Gt/year by 2030 (Sharifian et al., 2021).

2. Challenges in CO₂ capture methods

CO₂ capture is widely accepted as a vital procedure and presents many challenges (Gao et al., 2020; Gür, 2022). As mentioned before, the most important challenge in CO₂ capture is the large amount of energy consumption (Kumaravel et al., 2020). For example, in DAC, the energy and material costs of moving large amounts of air through an absorbent structure are also expected to result in high capture costs (Castel et al., 2021).

Separation of CO₂ from air, combustion exhausts or other process either from stationary point or distributed sources needs to be investigated and overcome obstacles (Gür, 2022). In general, CO₂ capture methods can be classified into chemical and physical categories. Tables 2 and 3 show some challenges in these methods.

Table 2
Challenges in chemical CO₂ capture.

Capture method		Challenges
Chemical looping	Oxy fuel	<ul style="list-style-type: none"> • High oxygen production energy costs • High sensitivity to air leakage into the system • Although this is an efficient capture method, it is difficult to retrofit as compared to the post-combustion method. • Special materials are needed to resist the high flame temperature (ca. 3500 °C). However, the recycled CO₂ can be used to moderate this temperature (Sharifian et al., 2021).
	Calcium (Ca) looping	<ul style="list-style-type: none"> • Rapid decrease of limestone (i.e., sorbent) capacity after several cycles of reaction with CO₂ • Environmental concerns due to limestone mining, waste from Ca-looping (i.e., spent calcium oxide (CaO)) and the need for high temperatures for operation • Need for air separation unit to obtain pure oxygen (O₂) for calcination (Sharifian et al., 2021).
Post combustion	Absorption	<ul style="list-style-type: none"> • High energy requirements (Wilberforce et al., 2021) • The limited CO₂ absorbing capacity is caused by the reaction stoichiometry and depends on the absorbent type (Sharifian et al., 2021).
	Amine	<ul style="list-style-type: none"> • Requires a regeneration step (Castel et al., 2021) • Low CO₂ loading capacity of the solvent • Amine degradation in the presence of nitrogen oxides (NO_x), sulfur oxides (SO_x), O₂, and particulate matter • Corrosion in amine equipment causes absorption column and stripper components to degrade over time. • The solvent cannot be completely regenerated. • The recovery step is energy intensive and the waste stream can be hazardous (Spigarelli and Kawatra, 2013).
	Ammonia	<ul style="list-style-type: none"> • The flue gas must be cooled to 15–27 °C due to volatility of ammonia. • High losses of ammonia vapor during stripping (Spigarelli and Kawatra, 2013)
	Adsorption	<ul style="list-style-type: none"> • Adsorbent degradation in cyclic operation • Possible decrease in adsorption capacity of the adsorbent after the desorption step (Sharifian et al., 2021) • Cyclic process requires regeneration (Castel et al., 2021).
	Activated carbons	<ul style="list-style-type: none"> • Low CO₂ capacity in mild conditions • The wide variety of raw materials means that a wide range of pore characteristics is often seen between adsorbents. • Negatively affected by NO_x, SO_x, and water (H₂O) (Spigarelli and Kawatra, 2013)
	Amine functionalized	<ul style="list-style-type: none"> • Degrades at a temperature of about 100 °C • Irreversible reactions with NO_x and SO_x produce unwanted byproducts • A temperature swing approach is required for desorption (Spigarelli and Kawatra, 2013).
Absorption	Solvay process	Dual alkali <ul style="list-style-type: none"> • High energy demand of calciner and CO₂ production from calcination for large-scale CO₂ capture (Spigarelli and Kawatra, 2013)
CO ₂ reduction technology (Albo et al., 2021)	Electrochemical reduction (Albo et al., 2021)	<ul style="list-style-type: none"> • Improved and/or novel electrocatalytic materials • Improved/novel electrolytes • More efficient electrocatalytic reactors (Albo et al., 2021)
	High-temperature molten carbonate cells	Difficult operating conditions due to: <ul style="list-style-type: none"> • High temperatures • Corrosion • Sensitivity to the presence of SO_x in the gas mixture (Sharifian et al., 2021)
	Redox active carriers and electrode reactions	<ul style="list-style-type: none"> • Limitation in terms of both solvents and carriers • Difficult finding a solvent that is inexpensive, safe and electrochemically stable • Difficult to allow high solubility of redox species (Sharifian et al., 2021)

Table 2
Continued.

	Proton coupled electron transfer (PCET) active agents	<ul style="list-style-type: none"> • Slow electrode kinetics • Low solubility of PCET organics • The sensitivity of the process to the impurities in the flue gas such as O₂, water and sulfur (Sharifian et al., 2021)
	Photochemical reduction	<ul style="list-style-type: none"> • Innovative photoactive materials • Efficient photocatalytic reactors (Albo et al., 2021)
	Photoelectrochemical reduction	<ul style="list-style-type: none"> • Development of photoactive materials • Efficient photocatalytic reactors (Albo et al., 2021)

Table 3
Challenges in physical CO₂ capture.

Situation	Capture method	Technology	Challenges
Post-combustion	Membrane		<ul style="list-style-type: none"> • Low concentration of CO₂ in fuel gas leads to higher energy consumption (Wilberforce et al., 2021). • Sensitivity to moisture (i.e., lower selectivity) (Sharifian et al., 2021) • The high temperature of flue gas degrades organic membranes. The gas must be cooled below 100 °C. • Membranes must be resistant to flue gas impurities, aging and plasticization (hardening). • Single-stage membrane systems are not capable of high CO₂ capture efficiency; a second stage is required (Spigarelli and Kawatra, 2013). • The low concentration of CO₂ in the fuel gas leads to high separation energy and the need for membranes with high selectivity. Therefore, it is not economical in terms of scale (Castel et al., 2021; Wilberforce et al., 2021; Spigarelli and Kawatra, 2013). • Trade-off between permeability and selectivity in polymeric membranes (Sandru et al., 2022) • Ion exchange extra energy and membranes require thermal stability is a limitation at high temperature (Sharifian et al., 2021).
	Adsorption	Zeolite	<ul style="list-style-type: none"> • The presence of impurities (NO_x, SO_x and H₂O) significantly affects performance. • It is time and energy consuming for complete regeneration as a temperature swing approach is required (Spigarelli and Kawatra, 2013).
		Metal-organic frameworks (MOFs)	<ul style="list-style-type: none"> • Negatively affected by NO_x, SO_x, and H₂O • Low CO₂ selectivity in carbon dioxide/nitrogen (CO₂/N₂) gas streams • Lack of experimental data on performance after multiple adsorption/desorption cycles • Pressure and temperature swing desorption approaches have not been adequately investigated (Spigarelli and Kawatra, 2013). • By increasing the porosity of MOFs, their mechanical stability is compromised. • The instability of MOFs may cause a phase change in them. • The cost of CO₂ capture through MOFs is high (Younas et al., 2020).
	Cryogenic distillation		<ul style="list-style-type: none"> • Very high energy requirement for DAC • The water content of the feed stream must be removed to avoid equipment clogging due to ice formation. • Buildup of solid CO₂ reduces the efficiency of the evaporator over time. • High capital cost of equipment • High cost of the refrigerant used to cool the system (Spigarelli and Kawatra, 2013)
Pre-combustion	Absorption	Selexol process	<ul style="list-style-type: none"> • Process is most efficient at elevated pressures (Spigarelli and Kawatra, 2013).
		Rectisol process	<ul style="list-style-type: none"> • The solvent is capable of absorbing trace metal compounds, such as mercury to form amalgams at low operating temperatures. • Solvent cooling leads to high operating costs. • Complex operating scheme leads to high capital costs (Spigarelli and Kawatra, 2013).
		Fluor process	<ul style="list-style-type: none"> • High solvent circulation rates (increasing operating cost) • Solvent cost (Spigarelli and Kawatra, 2013)

Table 3
Continued.

Purisol process	<ul style="list-style-type: none"> • Additional compression is required after the water-gas shift reaction (Spigarelli and Kawatra, 2013). • Retrofitting of existing plants is expensive and more difficult compared to oxy-fuel and post-combustion. • Syngas must be dried before CO₂ capture. • For non-gaseous feed stocks (e.g., coal or crude oil) the syngas stream must be cleaned of impurities in the gasification material. • The integrated gasification combined cycle (IGCC) system has high investment and operational costs (Spigarelli and Kawatra, 2013).
-----------------	--

Note that many of the physical solvents used in pre-combustion CO₂ capture can be used in post-combustion CO₂ capture if the gas stream is properly clean and pressurized, so the challenges are the same (Spigarelli and Kawatra, 2013).

3. Challenges in CO₂ conversion methods

3.1. Challenges in conventional CO₂ conversion methods

The production of synthetic hydrocarbon fuels and chemicals often requires high purity CO₂ feeds (Sharifian et al., 2021). In these carbon-based fuels, CO₂ is ultimately released into the atmosphere elsewhere in the value chain (Galán-Martín et al., 2022).

3.1.1. Membrane technology

The increased purity (selectivity) obtained by the high-performance membrane material is associated with significantly lower productivity (permeability) (Ebadi Amooghin et al., 2019; Ebadi Amooghin et al., 2022; Bandeali et al., 2021; Mashhadikhan et al., 2021; Nematollahi et al., 2022). The increase in selectivity in fact systematically induces the need for a larger membrane surface area due to the faster decrease in driving force with increasing permeate purity (Castel et al., 2021).

3.1.2. Sorbent

In high-temperature solid sorbents for CO₂ capture (in pre-combustion applications), several challenges arise at the adsorbent, reactor, and system scales (Gao et al., 2020).

3.1.3. Heterogeneous catalyst

For heterogeneous catalyst, there are important challenges related to the development of practical catalytic systems for CO₂ utilization. These challenges include the limited range of products due to their thermodynamic stability, the large energy barriers due to their kinetic stability, and the restraining by the produced water of the hydrogenation reaction (Burkart et al., 2019).

3.2. Challenges in CO₂ conversion through reduction methods

The major problem with many systems using this technology is that only a small amount of CO₂ (< 20%) is converted to product and the CO₂ reacts with hydroxide (OH⁻) in the electrolyte to form carbonates (Burkart et al., 2019).

3.2.1. Photoelectrochemical technology

The following challenges must be addressed for large-scale production through photoelectrochemical (PEC) CO₂ conversion (Kumaravel et al., 2020):

- *Stability*

Some inexpensive electrode materials are unstable and scale up procedures are expensive.

- *Light absorption*

Some additional studies are needed for precious metals to improve selectivity and product efficiency.

- *Yield and selectivity*

1. The main issue is bond-breaking/making reactions at the electrode–electrolyte interface.

2. Ionic liquids are one of the promising electrolytes with high product selectivity, but they are very expensive.

3.3. Hybrid methods

3.3.1. MOF-Catalyst

- For direct solid-gas phase conversion of CO₂, the selectivity of the MOF-based catalyst should be further improved because water and N₂ can compete with CO₂.

- The instability of MOFs in the presence of SO_x and NO_x (Younas et al., 2020).

3.3.2. Photo-Catalyst

Designing a photo-catalyst active in visible light and at a same time having favorable properties for the selective reduction of CO₂ remains a challenge (Gao et al., 2020).

4. Concluding remarks

Grand challenges in CO₂ capture and conversion were investigated. Challenges in chemical and physical CO₂ capture methods were discussed and summarized in Tables 2 and 3. Moreover, conventional, advanced, and hybrid CO₂ conversion methods and the related challenges were studied. In general, the four main challenges in CO₂ capture and conversion that need to be solved are: (1) the energy consumption of existing CO₂ capture technologies, especially at low CO₂ concentrations, (2) fixed and operating costs, (3) the need to development of highly active, stable, and economical catalysts or microorganisms for CO₂ utilization, and (4) carbon footprint in carbon capture and utilization technologies.

Along with the research in the development of existing technologies for CO₂ capture and conversion, as a perspective for future studies, the life cycle assessment and value chain analysis in the separation, reuse or conversion of CO₂ as a carbon source, need further investigations. It is recommended to further study the engineering aspects of the carbon footprint of CO₂ capture and conversion technologies to explore creative solutions to the issue.

Abbreviations	Definition
Ca looping	Calcium looping
CaO	Calcium oxide
CCU	Carbon capture and utilization
CO ₂	Carbon dioxide
DAC	Direct air capture
GHG	Greenhouse gas
Gt/year	Gigatonnes per year
H ₂ O	Water
IGCC	Integrated gasification combined cycle
MOFs	Metal organic frameworks
N ₂	Nitrogen
NO _x	Nitrogen oxides
O ₂	Oxygen
OH ⁻	Hydroxide
PCET	Proton coupled electron transfer
PEC	Photoelectrochemical
SO _x	Sulfur oxides

References

- Albo, J., Alvarez-Guerra, M., Irabien, A., 2021. Electro-, photo-, and photoelectrochemical reduction of CO₂, in: Teoh, W.Y., Urakawa, A., Ng, Y.H., Sit, P. (Eds.), *Heterogeneous catalysts: Advanced design, characterization and applications*. Vol. 1, Wiley-VCH GmbH, Weinheim, Germany, pp. 649-669. <https://doi.org/10.1002/9783527813599.ch36>
- Bandehali, S., Ebadi Amooghin, A., Sanaeepur, H., Ahmadi, R., Fuoco, A., Jansen, J.C., Shirazian, S., 2021. Polymers of intrinsic microporosity and thermally rearranged polymer membranes for highly efficient gas separation. *Sep. Purif. Technol.* 278, 119513. <https://doi.org/10.1016/j.seppur.2021.119513>
- Burkart, M.D., Hazari, N., Tway, C.L., Zeitler, E.L., 2019. Opportunities and challenges for catalysis in carbon dioxide utilization. *ACS Catal.* 9, 7937-7956. <https://doi.org/10.1021/acscatal.9b02113>
- Castel, C., Bounaceur, R., Favre, E., 2021. Membrane processes for direct carbon dioxide capture from air: Possibilities and limitations. *Front. Chem. Eng.* 3, 668867. <https://doi.org/10.3389/fceng.2021.668867>
- Centi, G., Perathoner, S., 2023. The chemical engineering aspects of CO₂ capture, combined with its utilization. *Curr. Opin. Chem. Eng.* 39, 100879. <https://doi.org/10.1016/j.coche.2022.100879>
- Chung, W., Lim, H., Lee, J.S., Al-Hunaidy, A.S., Imran, H., Jamal, A., Roh, K., Lee, J.H., 2022. Computer-aided identification and evaluation of technologies for sustainable carbon capture and utilization using a superstructure approach. *J. CO₂ Util.* 61, 102032. <https://doi.org/10.1016/j.jcou.2022.102032>
- De Ras, K., Van de Vijver, R., Galvita, V.V., Marin, G.B., Van Geem, K.M., 2019. Carbon capture and utilization in the steel industry: challenges and opportunities for chemical engineering. *Curr. Opin. Chem. Eng.* 26, 81-87. <https://doi.org/10.1016/j.coche.2019.09.001>
- Ebadi Amooghin, A., Mashhadikhan, S., Sanaeepur, H., Moghadassi, A., Matsuura, T., Ramakrishna, S., 2019. Substantial breakthroughs on function-led design of advanced materials used in mixed matrix membranes (MMMs): A new horizon for efficient CO₂ separation. *Prog. Mater. Sci.* 102, 222-295. <https://doi.org/10.1016/j.pmatsci.2018.11.002>
- Ebadi Amooghin, A., Sanaeepur, H., Luque, R., Garcia, H., Chen, B., 2022. Fluorinated metal-organic frameworks for gas separation. *Chem. Soc. Rev.* 51, 7427-7508. <https://doi.org/10.1039/D2CS00442A>
- Galán-Martín, Á., del Mar Contreras, M., Romero, I., Ruiz, E., Bueno-Rodríguez, S., Eliche-Quesada, D., Castro-Galiano, E., 2022. The potential role of olive groves to deliver carbon dioxide removal in a carbon-neutral Europe: Opportunities and challenges. *Renew. Sust. Energ. Rev.* 165, 112609. <https://doi.org/10.1016/j.rser.2022.112609>
- Gao, W., Liang, S., Wang, R., Jiang, Q., Zhang, Y., Zheng, Q., Xie, B., Toe, C.Y., Zhu, X., Wang, J., Huang, L., 2020. Industrial carbon dioxide capture and utilization: state of the art and future challenges. *Chem. Soc. Rev.* 49, 8584-8686. <https://doi.org/10.1039/D0CS00025F>
- Gür, T.M., 2022. Carbon dioxide emissions, capture, storage and utilization: Review of materials, processes and technologies. *Prog. Energy Combust. Sci.* 89, 100965. <https://doi.org/10.1016/j.pecs.2021.100965>
- Kumaravel, V., Bartlett, J., Pillai, S.C., 2020. Photoelectrochemical conversion of carbon dioxide (CO₂) into fuels and value-added products. *ACS Energy Lett.* 5, 486-519. <https://doi.org/10.1021/acseenergylett.9b02585>
- Li, W., Wang, H., Jiang, X., Zhu, J., Liu, Z., Guo, X., Song, C., 2018. A short review of recent advances in CO₂ hydrogenation to hydrocarbons over heterogeneous catalysts. *RSC Adv.* 8, 7651-7669. <https://doi.org/10.1039/C7RA13546G>
- Mashhadikhan, S., Ebadi Amooghin, A., Moghadassi, A., Sanaeepur, H., 2021. Functionalized filler/synthesized 6FDA-Durene high performance mixed matrix membrane for CO₂ separation. *J. Ind. Eng. Chem.* 93, 482-494. <https://doi.org/10.1016/j.jiec.2020.10.033>
- Nematollahi, K., Salehi, E., Ebadi Amooghin, A., Sanaeepur, H., 2022. CO₂ separation of a novel Ultem-based mixed matrix membrane incorporated with Ni²⁺-exchanged zeolite X. *Greenh. Gases: Sci. Technol.* 12, 48-66. <https://doi.org/10.1002/ghg.2122>
- Sandru, M., Sandru, E.M., Ingram, W.F., Deng, J., Stenstad, P.M., Deng, L., Spontak, R.J., 2022. An integrated materials approach to ultrapermeable and ultraselective CO₂ polymer membranes. *Science.* 376, 90-94. <https://doi.org/10.1126/science.abj9351>
- Sharifian, R., Wagterveld, R.M., Digdaya, I.A., Xiang, C., Vermaas, D.A., 2021. Electrochemical carbon dioxide capture to close the carbon cycle. *Energy Environ. Sci.* 14, 781-814. <https://doi.org/10.1039/D0EE03382K>
- Spigarelli, B.P., Kawatra, S.K., 2013. Opportunities and challenges in carbon dioxide capture. *J. CO₂ Util.* 1, 69-87. <https://doi.org/10.1016/j.jcou.2013.03.002>
- U.S. Energy Information Administration, 2021. U.S. energy-related carbon dioxide emissions. 2021. <https://www.eia.gov/environment/data>
- Valluri, S., Claremboux, V., Kawatra, S., 2022. Opportunities and challenges in CO₂ utilization. *J. Environ. Sci.* 113, 322-344. <https://doi.org/10.1016/j.jes.2021.05.043>
- Wei, K., Guan, H., Luo, Q., He, J., and S. Sun, 2022. Recent advances in CO₂ capture and reduction. *Nanoscale.* 2022. 14: 11869-11891. <https://doi.org/10.1039/D2NR02894H>
- Wilberforce, T., Olabi, A.G., Sayed, E.T., Elsaid, K., Abdelkareem, M.A., 2021. Progress in carbon capture technologies. *Sci. Tot. Environ.* 761, p.143203. <https://doi.org/10.1016/j.scitotenv.2020.143203>
- Younas, M., Rezakazemi, M., Daud, M., Wazir, M.B., Ahmad, S., Ullah, N., Ramakrishna, S., 2020. Recent progress and remaining challenges in post-combustion CO₂ capture using metal-organic frameworks (MOFs). *Prog. Energy Combust. Sci.* 80, 100849. <https://doi.org/10.1016/j.pecs.2020.100849>